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in this set (55) what is
a chemical bond? a
mutual electrical
attraction between
the nuclei and
valence electron of
different atoms that
binds the atoms
together.

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Choice Identify the choice that best completes the

statement or answers the question. Put the LETTER of the correct answer in the blank. _____ 1. The electrons involved in the formation of a chemical bond are called a. dipoles. c. Lewis electrons. b. s

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electrons. d. valence
electrons. _____ 2.

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bonding. a mutual

electrical attraction

between the nuclei

and the valence

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electrons of different atoms that binds the atoms together. ionic bonding.

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Chemical Bonding
SECTION 5 SHORT
ANSWER Answer the
following questions in
the space provided. 1.

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Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?

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molar mass of C_6H_6O

= 94.11 g 21. a. molar

mass of SO_3 = 80.07

g 49.2 mg SO_3 1 g

1000 mg 1 mol 80.07

g = 6.14×10^{-4} mol SO

3 b. molar mass of

PbO_2 = 239.2 g 7.44

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x 10⁴ kg PbO 2 1000
g 1 kg 1 mol 239.2 g
= 3.11 10⁵ mol PbO 2
c. molar mass of
CHCl₃ = 119.37 g
59.1 g CHCl₃ 1 mol
119.37 g = 0.495 mol
CHCl₃ d. molar mass
of C₂H₃Cl₃ = 133.39
g 3.27 ...

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discover a extra
experience and
execution by
spending more cash.
nevertheless when?
realize you tolerate
that you require to
acquire those every
needs in the manner

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of having significantly
cash? Why don't you
attempt to acquire
something basic in
the beginning?

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?questionalkyl halide

answerhalogen atom

bonded to an sp^3

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hybridized carbon
atom that is
tetrahedral
question primary alkyl
halide ...

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Statement that when
the elements are
arranged by

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number, there is a
periodic repetition of
their chemical and
physical properties. ...

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Bonding | Covalent
Bond. Chapter 6 -
Answer Key to
Section Review 1-3.
Section Review 1 1.
What is the main
distinction between
ionic and covalent

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bonding? Answer (A):
Ionic bonding involves
the electrical
attraction between
large numbers of
anions and cations.
Covalent bonding
involves the sharing
of electron pairs... <https://www.scribd.com/document/173653258/Chemistry-Chapter-6-Review>

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hazardous material is
defined by DOT as a.
a material that
possesses a threat to
the health and safety
of employees in the
work b. any material

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Considered harmful or
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the environment.

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with the wavelike

properties of light:

frequency (ν),

wavelength (λ), and

energy (E) as well as

the equations that

show their

relationships ($E = h$

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